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Primary amyl, hexyl, octyl and phenylpropyl alcohols
form a mixture of primary and secondary bromides
when acted upon by sodium bromide and sulfuric acid.
The mixture of bromides results, apparently, from
migration of the bromine atom from α to β position
under the catalytic action of sodium ions. When
hydrogen bromide is used instead of sodium bromide,
no migration of this type is observed. In the Wurtz
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